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First-principles study of crystal structural stability and electronic and magnetic properties in LaMn₇O₁₂

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Abstract

The crystal structure, electronic and magnetic properties of $LaMn_7O_{12}$ (($LaMn_3^{3+}$)_A $Mn_4^{3+}O_{12}$) are investigated by GGA (LSDA) and GGA + U (LSDA + U) ($0.0 \le U \le 5.0 \text{ eV}$) methods. Based on two experimentally refined structures (distinguished by the distortion parameter Δ , namely \mathbf{S}_{I} ($\Delta = 8.5 \times 10^{-5}$) and \mathbf{S}_{II} ($\Delta = 25.0 \times 10^{-4}$)), GGA and GGA + U with U < 3.0 eV calculations indicate that S₁ with a small distortion is the lowest-energy crystal structure while GGA + U with $3.0 \le U \le 5.0$ eV calculations show that S_{II} with a larger distortion is the ground-state crystal structure. Within the LSDA method, S_{II} is always the ground-state structure no matter if U is considered or not. There are two independent magnetic sublattices: Mn³⁺ within the A site and Mn³⁺ within the B site. First, it is predicted that A-site Mn³⁺ ions are preferably AFM-coupled in G-type (antiferromagnetically coupled in three directions). Based on this result, four magnetic configurations (**FM**-A^{\uparrow}B^{$\uparrow\uparrow$}, **AFM1**-A^{\uparrow}B^{$\downarrow\downarrow$}, **AFM2**- $A^{\uparrow\downarrow}B^{\uparrow\uparrow}$ and **AFM3**- $A^{\uparrow\downarrow}B^{\uparrow\downarrow}$) are designed, and their total energies are calculated. Our results demonstrate that AFM2 and AFM3 are the lowest magnetic state, respectively, for S_{I} and S_{II} . Correspondingly, LaMn₇O₁₂ is metallic with no orbital ordering at AFM2 for S_{I} while it is an insulator with orbital ordering at AFM3 for S_{II} . Thus, modulation of the distortion parameter Δ , e.g. by chemical doping, could be employed as a new avenue to induce a magnetic phase transition and the corresponding metal-to-insulator transition in LaMn₇O₁₂.

(Some figures in this article are in colour only in the electronic version)

1. Introduction

Spin, charge, orbital and lattice degrees of freedom play important roles in the electronic, magnetic and transport properties of transition metal oxides [1-6]. On explaining these couplings, Goodenough [7] showed that the dominant

role of the Anderson superexchange interaction [8] between neighboring Mn ions was through the Mn–O–Mn path of the corner-sharing network of MnO₆ octahedra. The Goodenough–Kanamori–Anderson (GKA) rules [7, 9, 10] provide a model to describe the strength and orientation of the magnetic interaction, in which the key parameters are the Mn–O–Mn bond angles and the Mn–O bond length. It is demonstrated that the stabilization of a given charge, magnetic and orbital orderings requires a well-defined pattern of the cooperative buckling and distortion of the octahedral network.

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In order to validate the GKA rules, Prodi *et al* synthesized single crystals of LaMn₇O₁₂ under high pressure at P = 4 GPa and high temperature T = 1000 °C [11]. LaMn₇O₁₂ belongs to the A'A₃B₄O₁₂-type double perovskites, which have received wide attention owing to their special ordered structures and a variety of physical properties [12–18]. In LaMn₇O₁₂ there are only Mn³⁺, thus only magnetic and orbital orderings need to be considered, excluding the charge ordering effect. Their crystal structure is refined to the monoclinic I2/m space group. As for their magnetic property, it is found that there exist two transition temperatures: $T_{\rm N,B} = 78$ K due to the C-type antiferromagnetic (AFM) structure within B-site Mn³⁺ ions and $T_{\rm N,A} = 21$ K due to a second AFM structure within Asite Mn³⁺ ions.

Actually in 1974, Bochu *et al* also successfully refined the structure of $LaMn_7O_{12}$ in the monoclinic I2/m space group [19]. Detailed comparison between these two experimentally refined structures reveals that they show remarkable differences, especially for the bond length differences around the two B-site Mn ions (which will be shown below). Since it is believed that in $LaMn_7O_{12}$ the existence of distinct Jahn–Teller (JT) B-site Mn ions plays a primary role in determining the magnetic coupling, we suspect that different distortions around the B-site Mn ions should quantitatively, and even qualitatively, influence their magnetic properties and then the corresponding electronic properties.

It is well known that in transition metal oxides strong local Coulomb interactions play very important roles, and are responsible for the observed ground states and the associated physical properties. For instance, the electronic interactions could suppress charge fluctuations in Mott insulators and lead to rather strong coupling among spin, charge, orbital and lattice degrees of freedom [20]. Thus the standard density-functional theory (DFT) often fails to reproduce the observed electronic structure and magnetic properties. Indeed, two distinct ground states with either checkerboard charge ordering or Zener polaron formation were reported, depending on the proportion of Hartree–Fock exchange used in $La_{0.5}Ca_{0.5}MnO_3$ [21]. Similarly, the charge-orbital ordering and hence the Verwey metal–insulator transition was shown to be driven by the onsite Fe d-electron correlation in Fe₃O₄ [22].

Therefore, it becomes necessary that in the study of LaMn₇O₁₂ the following three key issues be resolved: (i) which is the most stable structure for the two experimentally refined I2/m crystal structures; (ii) how the different Jahn–Teller distortions within B-site Mn³⁺ ions affect their magnetic interactions and electronic structures and (iii) how the Coulomb interaction affects its structural stability and electronic and magnetic properties. Besides our studies on these three issues, we will also present a detailed investigation on the magnetic coupling within A-site Mn³⁺ ions since in LaMn₇O₁₂ there exists a second sublattice of JT Mn³⁺ ions in A sites with a square coordination.

2. Crystal structure and magnetic orders

LaMn₇O₁₂ is refined in the monoclinic I2/m space group. It is a pseudocubic perovskite-like structure which belongs

Table 1. Crystal parameters and selected bond lengths (in angströms) for LaMn₇O₁₂ by Bochu *et al* [19] (**S**_I) and Prodi *et al* [11] (**S**_{II}). Δ is used to characterize the distortion which is defined as $\Delta = (1/N) \sum_{n=1,N} \{(d_n - \langle d \rangle)/\langle d \rangle\}^2$. The energy differences (meV) between different magnetic configurations taking **FM** as reference obtained using GGA + U with U equal to 3.0 and 4.0 eV.

	I2/m—S _I	$I2/m$ — S_{II}
(Crystal parameters	
a (Å)	7.516	7.509
b (Å)	7.376	7.349
<i>c</i> (Å)	7.516	7.504
β (deg)	91.30	91.354
	Bond length Å	
Mn _{4e} –O ₁	2.012	2.145
Mn _{4e} -O ₃	1.991	1.968
Mn _{4e} -O ₄	1.967	1.908
Δ_{4e}	$8.5 imes 10^{-5}$	25.0×10^{-4}
$Mn_{4f}-O_2$	1.990	1.890
$Mn_{4f}-O_3$	1.973	1.987
Mn _{4f} –O ₄	2.010	2.129
$\Delta_{ m 4f}$	5.8×10^{-5}	24.0×10^{-4}
GGA + U (3.	0 eV) calculated resu	ults (meV)
$A^{\uparrow\uparrow}B^{\uparrow\uparrow}$ (FM)	0.0	0.0
$A^{\uparrow\uparrow}B^{\downarrow\downarrow}$ (AFM1)	-34	-45
$A^{\uparrow\downarrow}B^{\uparrow\uparrow}$ A-AFN	412 -435	-211
(AFM2) G-AFM	M2 –474	-246
$A^{\uparrow\downarrow}B^{\uparrow\downarrow}$ (AFM3)	-260	-526
GGA + U (4)	0 eV) calculated resu	ults (meV)
$A^{\uparrow\uparrow}B^{\uparrow\uparrow}$ (FM)	0.0	0.0
$A^{\uparrow\uparrow}B^{\downarrow\downarrow}$ (AFM1)	-57	-59
$A^{\uparrow\downarrow}B^{\uparrow\uparrow}$ A-AFN	√12 −542	-239
	10	
(AFM2) G-AFM	M2 - 576	-275

to the family of quadruple perovskites with general formula $A'A_3B_4O_{12}$, where A' can be a nonmagnetic monovalent, divalent or trivalent cation, and A and B are the JT ions. In LaMn₇O₁₂, both the A and B sites are occupied by Mn^{3+} ions, resulting in two interpenetrating sublattices of JT ions with octahedral (B-site Mn^{3+}) and square-planar (A-site Mn^{3+}) coordination (figures 1(a) and (b)). Figure 1(b) shows that the monoclinic distortion splits the pristine eightfold B site into two fourfold 4e and 4f sites and the pristine sixfold A sites into three twofold sites (A1, A2 and A3) with 2/m point symmetry. The refined crystal parameters and Mn-O bond length of the B sites by Bochu [19] (designated as S_I) and Prodi [11] (designated as S_{II}) are listed in table 1. From table 1, we see that the bond length differences around 4e and 4f are larger in S_{II} than those in S_{I} . For example, in S_{II} the bond length differences are 0.25 Å and 0.23 Å around 4e and 4f, respectively, while they are only 0.04 Å and 0.03 Å in S_I . The distortion parameter Δ has been used to characterize the JT distortion around the JT ions [23] and it is defined as Δ = $(1/N) \sum_{n=1,N} \{ (d_n - \langle d \rangle) / \langle d \rangle \}^2$ in which $\langle d \rangle$ is the average Mn-O bond distance. Based on this formula, the distortion parameter (Δ) is found to be 5.8 \times 10⁻⁵ and 24.0 \times 10⁻⁴, respectively, for S_I and S_{II} , an order-of-magnitude difference.

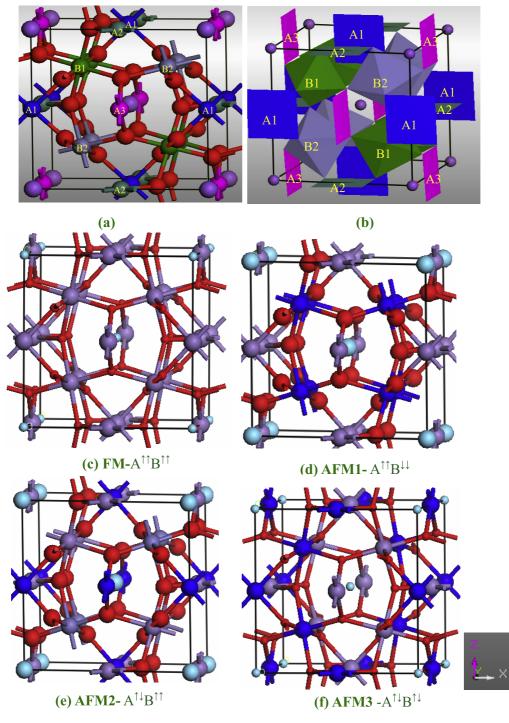


Figure 1. (a) Crystal structures of $LaMn_7O_{12}$ in I2/m monoclinic space group and (b) its monoclinic distortion splitting the B-site Mn ions into B1 (4e) and B2 (4f) sites and A-site Mn ions into A1, A2 and A3. (c)–(f) Present the four designed magnetic configurations (FM, AFM1, AFM2 and AFM3, respectively) with purple (gray) denoting the spin-up direction and blue (black) the spin-down direction.

Thus by comparative study of their magnetic and electronic properties on these two structures, the delicate interaction between the structural distortion and spin/orbital orderings in this compound can be investigated.

There exist two magnetic sublattices (A and B) in this compound. It was found experimentally that their orderings occurred independently from each other and that their magnetic structure of the B sublattice turned out to be of the C type (i.e. *antiferromagnetically* coupled to the *ferromagnetic ac*

plane). In order to study the magnetic couplings, *four* magnetic configurations are taken into account: (1) **FM** ($A^{\uparrow\uparrow}B^{\uparrow\uparrow}$) where the Mn ions are ferromagnetically coupled within A and B sites, respectively, and they are also ferromagnetically coupled between A and B sites (figure 1(c)); (2) **AFM1** ($A^{\uparrow\uparrow}B^{\downarrow\downarrow}$) where the Mn ions are ferromagnetically coupled within A and B sites, respectively, but they are antiferromagnetically coupled between A and B sites (figure 1(d)); (3) **AFM2** ($A^{\uparrow\downarrow}B^{\downarrow\downarrow}$) where the Mn ions are antiferromagnetically coupled between A and B sites (figure 1(d)); (3) **AFM2** ($A^{\uparrow\downarrow}B^{\downarrow\downarrow}$) where the Mn ions are antiferromagnetically coupled between A and B sites (figure 1(d)); (3) **AFM2** ($A^{\uparrow\downarrow}B^{\downarrow\downarrow}$) where the Mn ions are antiferromagnetically coupled between A mile Mn ions are antiferromagnetically coupled mn is a mile Mn ions are antiferromagnetically

within A sites and ferromagnetically coupled within B sites (figure 1(e)) and (4) **AFM3** ($A^{\uparrow\downarrow}B^{\uparrow\downarrow}$) where the Mn ions are antiferromagnetically coupled within A and B sites, respectively, and also antiferromagnetically coupled between A and B sites (figure 1(f)). In addition, in order to find which AFM configuration has the lowest energy within A-site Mn³⁺, based on the designed **AFM2** configuration, we arranged the A-site Mn³⁺ in A-type AFM (designated as **A-AFM2**), as tentatively determined experimentally [11] and G-type AFM (designated as **G-AFM2**) as determined from its isostructural compound NaMn₇O₁₂ [24].

3. Computational details

All calculations in this paper are carried out using the highly accurate full-potential linearized augmented plane wave plus local orbitals (FP-LAPW + lo) method [25, 26] with density-functional theory implemented in the WIEN2K package [27, 28]. In this method, the space is divided into an interstitial region and the nonoverlapping muffin-tin (MT) spheres are centered at the atomic sites. In the MT region, the basis sets are described by radial solutions of the one-particle Schrödinger equation (at fixed energy) and their energy derivatives are multiplied by spherical harmonics.

In order to achieve energy convergence, the wavefunctions in the interstitial region are expanded in plane waves with a cutoff $R_{\rm MT}^{\rm min} K_{\rm max} = 7$, where $R_{\rm MT}^{\rm min}$ denotes the smallest atomic sphere radius and K_{max} the magnitude of the largest K vector in the plane wave expansion. The values of the atomic sphere radii (R_{MT}) are taken as 2.50, 1.88 and 1.68 au for La, Mn and O, respectively. For the electron population analysis, the muffin-tin radii $(R_{\rm MT})$ are chosen to be the touching spheres to keep as much core charges inside the muffin tin as possible. The valence wavefunction inside the spheres is expanded up to $l_{\text{max}} = 10$, while the charge density in terms of the Fourier series is expanded up to $G_{\text{max}} = 12$. The total Brillouin zone is sampled with 1000 k-points. The Brillouin zone integration is carried out with a modified tetrahedron method [29]. The self-consistent field is considered to be achieved when the total energy difference between succeeding iterations is less than 10^{-5} Ryd.

As for the exchange–correlation potential, we employ both the standard generalized gradient approximation (GGA) using the Perdew–Burke–Ernzerhof scheme [30] and the local spin density approximation (LSDA) using Cerply-Alder. In addition, to properly describe the strong electron correlation in the Mn³⁺ 3d orbitals, the GGA plus on-site repulsion U method (GGA + U and LSDA + U) is also used [31–33] with effective U values ($U_{\text{eff}} = U - J$) of 0.0–5.0 eV for Mn. Unless otherwise indicated, the reported results are based on GGA + U_{eff} (with $U_{\text{eff}} = 4.0$ eV for Mn).

4. Results and analyses

4.1. Magnetic coupling within A-site Mn³⁺

We first consider the magnetic structure of Mn^{3+} within the A sublattice. The topology of this sublattice differs from

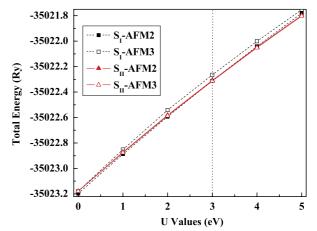


Figure 2. Variation of the total energies of S_I and S_{II} at AFM2 and AFM3 magnetic configurations versus different U values.

that of the B sublattice, as seen from figure 1(b). The nearest neighbors of each Mn are four, instead of six, and form a square within the plane of the ions, instead of an According to the neutron data, Prodi et al octahedron. tentatively designated the A-type AFM configuration within this sublattice [11]. However, we noticed that, in $NaMn_7O_{12}$, which is also in the same I2/m space group, the A-site Mn ions were believed to be the G-type AFM [24]. Hence, we calculated the energies of both A-AFM2 and G-AFM2 as defined above in which the A-site Mn^{3+} ions are in an A-type and G-type AFM arrangement, respectively. It is observed from table 1 that for both S_I and S_{II} , the energy of the Gtype AFM arrangement is always lower by about 35 meV than that of the A-type. Thus in the following analysis, the AFM coupling of A-site Mn³⁺ is referred to the G-type AFM.

4.2. Electronic correlation effect on crystal structure and magnetic property

Figure 2 shows the variation of the total energies of AFM2 (solid) and AFM3 (empty) in both S_{I} (square) and S_{II} (triangle) structures versus the U values based on the GGA + U method. It is obvious that, when U < 3.0 eV, S_I is the lowestenergy crystal structure, while when $U \ge 3.0$ eV, S_{II} becomes the ground-state crystal structure. Similar results were also obtained in La_{0.5}Ca_{0.5}MnO₃ using the hybrid Hartree-Fock/density-functional theory *ab initio* method [21]. It was found that the ground-state charge ordering of La_{0.5}Ca_{0.5}MnO₃ could switch between checkerboard charge ordering and a Zener polaron one if the percentage of Hartree-Fock (HF) exchange was varied. These two states are therefore similar in energy and the actual ground state might depend on the specific A (La) and B (Mn) ions or even sample preparation, history and ambient conditions. In our calculation, an increase in U could be viewed as the increase of the percentage in HF exchange, and thus different ground-state crystal structures are also obtained with different U values. We consider that the different crystal structures obtained by Prodi et al [11] and Bochu et al [19] might be due to the different sample preparation and ambient conditions. Within the LSDA calculation, S_{II} is the ground-state crystal structure, and when

Table 2. LSDA calculated total energy differences (meV) between different magnetic configurations taking the FM state of S_I as a reference.

	$I2/m$ — S_I	$I2/m$ — \mathbf{S}_{II}
	LSDA	
$A^{\uparrow\uparrow}B^{\uparrow\uparrow}$ (FM)	0.0	195
$A^{\uparrow\uparrow}B^{\downarrow\downarrow}$ (AFM1)	-107	-235
$A^{\uparrow\downarrow}B^{\uparrow\uparrow}$ (AFM2)	-559	-421
$A^{\uparrow\downarrow}B^{\uparrow\downarrow}$ (AFM3)	-211	-596

the electronic correlation (U) is considered, S_{II} is more stable due to the larger energy difference between S_{I} and S_{II} .

We have also calculated the total energies for the four magnetic configurations of both S_{I} and S_{II} structures with results listed in table 1 based on GGA + U and in table 2 based on LSDA. It is clearly observed that, based on both GGA + U and LSDA, the two different structural distortions (S_{I} and S_{II}) lead to completely different magnetic interactions within B sites. Mn_{4e} and Mn_{4f} are likely to be AFM-coupled in S_{II} whilst they are favorable to be FM-coupled in S_{I} . Thus, the lowestenergy magnetic configurations is AFM2 for S_{I} , in which the B-site Mn ions are FM-coupled whilst the A-site Mn ions are AFM-coupled, and AFM3 for S_{II} , in which both B-site and Asite Mn ions are AFM-coupled. Our calculated prediction in S_{II} , where the two B-site Mn ions turn out to be AFM-coupled, is consistent with the experimental result by Prodi *et al* [11].

In figures 3(a) and (b), we present the total electronic density of states (TDOS) obtained by both the GGA (top row) and GGA + U (bottom row) methods for the AFM2 configuration in S_I and AFM3 in S_{II} , respectively. For S_I , both the GGA (LSDA) and GGA + U (LSDA + U) calculations predict a half-metallic character in its lowest-energy magnetic configuration AFM2, with a metallic character for the spin-up channel and semiconductor behavior for the spindown (figure 3(a)). For AFM3 configuration in S_{II} , GGA (LSDA) calculation shows a metallic character for LaMn₇O₁₂ (figure 3(b), top row), contrary to the insulator behavior experimentally determined. This indicates that the standard GGA (LSDA) calculation is unable to describe properly the electronic structure due to the strongly correlated nature of electrons in this material. The GGA + U (LSDA + U) method, on the other hand, is able to predict a clear insulator character, as shown in figure 3(b) in the bottom row with a gap of about 1.0 eV, compared to the experimental value of 0.22 eV. While the lower experimental bandgap could be due to the existence of midgap defect states which might be difficult to measure, our further calculation indicates that, with U varying from 1.0 to 5.0 eV, the energy gap changes from 0.20 to 3.0 eV. Therefore, a reliable measurement of the gap is needed in order to confirm the proper U values for this system.

In order to investigate the relationship between the magnetic coupling and electronic structure, we have also carried out the calculations for the **AFM3** magnetic configuration in S_I , with the result shown in figure 3(c). It is clear that LaMn₇O₁₂ also exhibits an insulator property in the assumed **AFM3** magnetic state in the S_I structure using the GGA + U (LSDA + U) method. These results indicate

that, in $LaMn_7O_{12}$, its local structure distortion within B-site Mn ions and the magnetic coupling behavior (AFM or FM) are delicately correlated. External pressure and/or chemical doping, which will change the distortion within B-site ions (e.g. substitution of La by other rare earth elements), could induce a magnetic phase transition within B-site Mn ions.

4.3. Electronic structure analysis

We then investigate the relationship between the magnetic property and electronic structures in this compound based on the GGA + U calculation. The site- and angular-projected partial density of states (PDOS) for Mn_{4e} and Mn_{4f} of AFM2 in S_I and AFM3 in S_{II} are shown in figures 4(a) and (b), respectively. The Mn^{3+} ion in $LaMn_7O_{12}$ is in a 3d⁴ configuration. In an octahedral crystal field, the Mn 3d orbitals will split into three lower t_{2g} and two upper e_g orbitals. However, because of the distorted octahedron as shown in figure 1(b) and table 1, the orbitals are further split. Thus three of the four electrons occupy the three t_{2g} orbitals whilst the remained fourth electron occupies the lower eg orbital. In LaMn₇O₁₂, the two B-site Mn^{3+} ions at the 4e and 4f sites possess distinct JT distorted octahedra, so the fourth electron should prefer different eg orbitals, resulting in orbital ordering in this compound. In more detail, in S_{II} as shown in figure 4(b), the t_{2g} orbitals (i.e. d_{xy} , d_{xz} and d_{yz}) in both Mn_{4e} and Mn_{4f} are occupied. Recalling the data in table 1 where the bond $Mn_{4e}-O_4$ (along the z direction) is the longest (2.1446 Å) and Mn_{4f}–O₄ is the shortest (1.8903 Å) for S_{II} , this distinct structural distortion would certainly lead to a different electron occupation for the fourth electron. The angular-PDOS shows that the fourth electron occupies the d_7^2 orbital on Mn_{4e} while it occupies the $d_{x^2-y^2}$ orbital on Mn_{4f}, which is consistent with their local structure distortion. For S_I , the site- and angular-PDOS, as plotted in figure 4(a) for its lowestenergy magnetic state AFM2, show that the fourth electron occupies almost equally the d_{z^2} and $d_{x^2-y^2}$ orbitals; thus there is no orbital ordering for S_I . In order to explicitly indicate different orbital orderings in these two different structures, figures 5(a) and (b) show the contour plots around the Fermi level. From these figures, one can clearly see that, for S_{II} in the AFM3 state, orbital ordering forms while for S_I in the AFM2 state, orbital ordering disappears. However, in the AFM3 magnetic configuration of S_I , similar PDOS analysis (figure 4(c)) shows that there also exists orbital ordering as observed in S_{II} , although their ordering is not very complete due to the comparatively small exchange split caused by a smaller distortion. These analyses indicate that, in LaMn₇O₁₂, magnetic coupling determines the existence of the orbital ordering, which is consistent with the GKA rules. In the GKA rules, the strength and orientation of the magnetic interaction are described and especially it provides the frame of the Jahn-Teller-driven polarization of the two relevant $e_g d_{x^2-y^2}$ and d_{z^2} orbitals in the MnO₆ octahedron. Comparison of the two structures S_I and S_{II} , their difference is mainly in the bond length differences (Δ). However, local structural distortion leads to different occupation $(d_{x^2-y^2} \text{ or } d_{z^2})$ of the fourth electron in the octahedral Mn ions, which is closely relevant to the magnetic ordering.

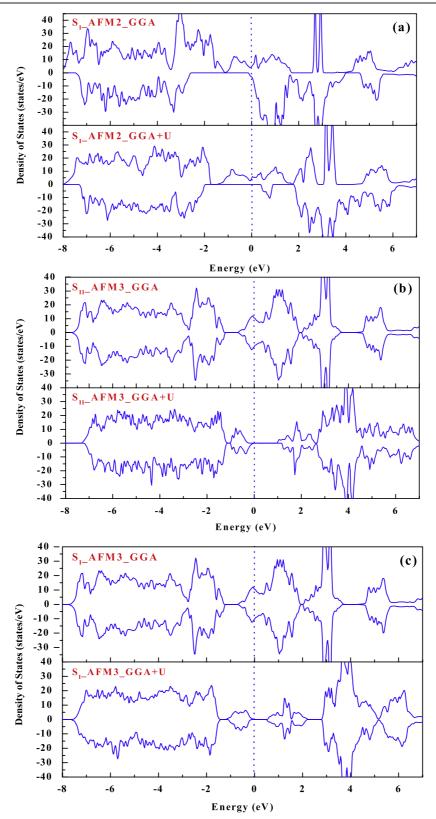


Figure 3. Total density of states for S_I in AFM2 magnetic state (a), S_{II} in AFM3 magnetic state (b) and S_I in AFM3 magnetic state (c) obtained by both GGA (top) and GGA + U (bottom) methods.

5. Conclusions

We have investigated the structural stability, electronic and magnetic properties of $LaMn_7O_{12}$ and the electronic

correlation effects on these properties by using GGA (LSDA) and GGA + U (LSDA + U) methods. Our results indicate clearly that the distortion (mainly the bond length difference) could significantly influence the magnetic and electronic

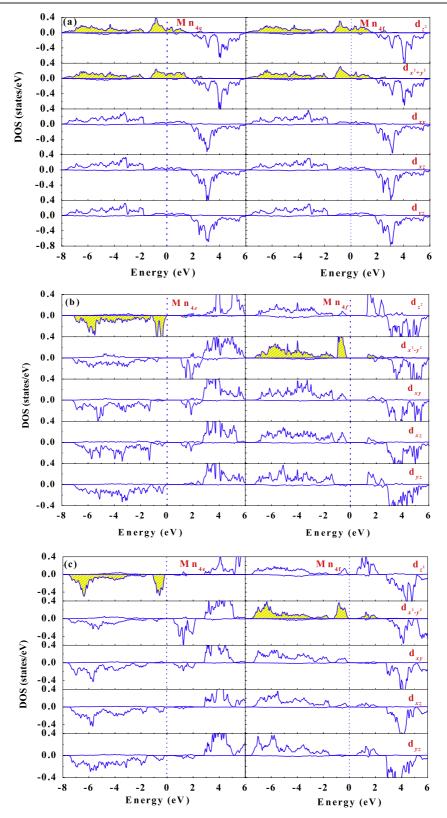


Figure 4. Partial density of states of the two B-site Mn ions (Mn_{4e} left and Mn_{4f} right) for S_I in the **AFM2** state (a), S_{II} in the **AFM3** state (b) and S_I in the **AFM3** state (c) obtained by the GGA + U ($U_{eff} = 4.0 \text{ eV}$) method.

properties and orbital ordering in the quadruple perovskite compound $LaMn_7O_{12}$. Total energy calculations for the four designed magnetic configurations of the two experimental structures show that in S_I , where Mn^{3+} ions have a small bond

length difference within B sites, the coupling is prone to be FM and that in S_{II} , where there is a large bond length difference within B-site Mn ions, the coupling is inclined to be AFM. Furthermore, the AFM coupling leads to an insulator behavior

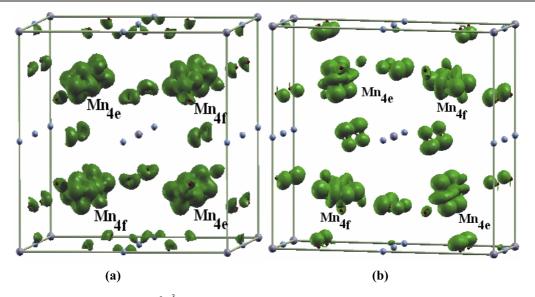


Figure 5. Spin density plot (isosurface at 0.3 e Å⁻³, mapped by XcrySDen) within the energy interval around -1.0-0.0 eV from GGA + U ($U_{\text{eff}} = 4.0$ eV) calculation in their lowest-energy magnetic states: S_I in AFM2 (a) and S_{II} in AFM3 (b).

and a clear orbital ordering of d_{z^2} and $d_{x^2-y^2}$ in the S_{II} structure while the FM coupling induces a metallic character with no orbital ordering in the S_I structure. For the A-sublattice Mn ions, the G-type AFM configuration is the lower energy state. Our GGA+U (LSDA+U) calculations further indicate that in LaMn₇O₁₂ the on-site Coulomb repulsion plays an important role in its crystal structure, electronic property and orbital ordering.

Acknowledgments

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References

- Kugel K I and Khomskii D I 1973 Sov. Phys.—JETP 37 725
 Kugel K I and Khomskii D I 1982 Sov. Phys.—Usp. 25 231
- [2] Tokura Y and Nagaosa N 2000 Science 288 462
- [3] Edwards A J and Peacock R D 1959 J. Chem. Soc. 4126 Scatturin V, Corliss L, Elliott N and Hastings J 1961 Acta Crystallogr. 14 19
- Yoneyama S and Hirakawa K 1966 *J. Phys. Soc. Japan* **21** 183 [4] Margadonna S and Karotsis G 2006 *J. Am. Chem. Soc.*
- **128** 16436 [5] Okimoto Y, Katsufuji T, Ishikawa T, Urushibara A,
- Arima T and Tokura Y 1995 Phys. Rev. Lett. 75 109
- [6] Radaelli P G, Cox D E, Marezio M and Cheong S W 1997 *Phys. Rev.* B 55 3015
- [7] Goodenough J B 1963 *Magnetism and Chemical Bond* (New York: Interscience)
- [8] Anderson P W 1959 Phys. Rev. 115 2
- [9] Kanamori J 1959 J. Phys. Chem. Solids 10 87
- [10] Anderson P W 1963 Solid State Physics vol 14, ed F Seits and D Tumbull (New York: Academic) p 99
- [11] Prodi A, Gilioli D, Cabassi R, Bolzoni F, Licci F, Huang Q Z, Lynn J W, Affronte M, Gauzzi A and Marezio M 2009 Phys. Rev. B 79 085105

- [12] Long Y W, Saito T, Tohyama T, Oka K, Azuma M and Shimakawa Y 2009 Inorg. Chem. 48 8489
- [13] Long Y W, Hayashi N, Saito T, Azuma M, Muranaka S and Shimakawa Y 2009 Nature 458 60
- [14] Shimakawa Y 2008 Inorg. Chem. 47 8562
- [15] Imamura N, Karppinen M, Motohashi T, Fu D S, Itoh M and Yamauchi H 2008 J. Am. Soc. Chem. 130 14948
- [16] Shiraki H, Saito T and Shimakawa Y 2008 Chem. Mater. 20 7077
- [17] Yamada I, Takata K, Hayashi N, Shinohara S, Azuma M, Mori S, Muranaka S, Shimakawa Y and Takano M 2008 Angew. Chem. Int. Edn 47 7032
- [18] Mizumaki M, Saito T, Shiraki H and Shimakawa Y 2009 Inorg. Chem. 48 3499
- [19] Bochu B, Chenavas J, Joubert J C and Marezio M 1974 J. Solid State Chem. 11 88
- [20] Imada M, Fujimori A and Tokura Y 1998 *Rev. Mod. Phys.* 70 1039
- [21] Patterson C H 2005 Phys. Rev. B 72 085125
- [22] Jeng H T, Guo G Y and Huang D J 2006 Phys. Rev. B 74 195115
- [23] Rodríguez-Carvajal J, Hennion M, Moussa F, Moudden A H, Pinsard L and Revcolevschi A 1998 Phys. Rev. B 57 R3189
- [24] Prodi A, Gilioli E, Gauzzi A, Licci F, Marezio F M, Bolzoni F, Huang Q, Santoro A and Lynn J W 2004 Nat. Mater. 3 48
- [25] Sjöstedt E, Nordström L and Singh D J 2000 Solid State Commun. 114 15
- [26] Madsen G K H, Blaha P, Schwarz K, Sjöstedt E and Nordström L 2001 Phys. Rev. B 64 195134
- [27] Schwarz K and Blaha P 2003 Comput. Mater. Sci. 28 259
- [28] Blaha P, Schwarz K, Madsen G K H, Kvasnicka D and Luitz L 2001 WIEN2k, An Augmented Plane Wave Plus Local Orbitals Program for Calculating Crystal Properties (Austria: Vienna University of Technology)
- [29] Blöchl P E 1994 *Phys. Rev.* B **50** 17953
- [30] Perdew J P, Burke K and Ernzerhof M 1996 Phys. Rev. Lett. 77 3865
- [31] Liechtenstein A I, Anisimov V I and Zaanen J 1995 Phys. Rev. B 52 R5467
- [32] Anisimov V, Aryasetiawa n F and Lichtenstein A 1997 J. Phys.: Condens. Matter 9 767
- [33] Petukhov A G, Mazin I I, Chioncel L and Lichtenstein A I 2003 Phys. Rev. B 67 153106